**SECTION 1 – IDENTIFICATION** 

Name, Address, and Telephone of the Responsible Party

Dyno Nobel Inc.

6440 S. Millrock Drive, Suite 150 Salt Lake City, Utah 84121

Phone: 801-364-4800 Fax 801-321-6703

E-Mail: dnna.hse@am.dynonobel.com www.dynonobel.com

Product Identifier: SDS 1022

**Product Name:** 

Ammonium Nitrate Solution 50-70%

Other Means of Identification

Synonyms:

Ammonium Nitrate Solution 50-70%

AN Solution

**Intended Use of the Product** 

Used in the manufacture of UAN and AN Prill

**Emergency Telephone Number** 

FOR 24 HOUR EMERGENCY, CALL CHEMTREC (USA) 800-424-9300

CANUTEC (CANADA) 613-996-6666

SECTION 2 – HAZARD(S) IDENTIFICATION

Classification of the Substance or Mixture

Classification (GHS-US)

Eye Irrit. 2B H320

Label Elements
GHS-US Labeling

Hazard Pictograms (GHS-US)

GHS07

Signal Word (GHS-US) : Warning

**Hazard Statements (GHS-US)** : H320 – Causes eye irritation.

**Precautionary Statements (GHS-**: P264 - Wash hands, forearms, and other exposed areas thoroughly after

US) handlii

P305+P351+P338 - IF IN EYES: Rinse cautiously with water for several

minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

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Date:

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P337+P313 - If eye irritation persists: Get medical advice/attention.

Other Hazards

Hazards Not Otherwise Classified (HNOC): Not available

SECTION 3 - COMPOSITION/INFORMATION ON INGREDIENTS					
Mixture					
Name	Product identifier	% (w/w)	Ingredient Classification (GHS-US)		
Water	CAS No) 7732-18-5	30-50	Not classified		



Ammonium nitrate	(CAS No) 6484-52-2	50-70	Ox. Sol. 3, H272
			Eye Irrit. 2B, H320

Ingredients, other than those mentioned above, as used in this product are not hazardous as defined under current Department of Labor regulations or are present in deminimus concentrations (less than 0.1% for carcinogens, less than 1.0% for other hazardous materials).

### **SECTION 4 - FIRST AID MEASURES**

#### **Description of First Aid Measures**

**General:** Never give anything by mouth to an unconscious person. If you feel unwell, seek medical advice (show the label where possible).

**Inhalation:** Allow breathing of fresh air. Allow the victim to rest.

**Skin Contact:** Remove affected clothing and wash all exposed skin area with mild soap and water, followed by warm water rinse.

**Eye Contact:** IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Ingestion: Rinse mouth. Do NOT induce vomiting. Obtain emergency medical attention.

#### Most Important Symptoms and Effects Both Acute and Delayed

Eye Contact: May cause eye irritation.

Indication of Any Immediate Medical Attention and Special Treatment Needed

No additional information available.

## **SECTION 5 - FIRE-FIGHTING MEASURES**

**Extinguishing Media** 

Suitable Extinguishing Media: Water spray.

**Unsuitable Extinguishing Media:** Use heavy water stream. **Special Hazards Arising from the Substance or Mixture** 

**Advice for Firefighters** 

**Precautionary Measures Fire:** Hot AN solution may also ignite combustibles such as wood, paper, oil, clothing, etc. It will support and increase the rate of combustion in the presence of flammable or combustible materials even in the absence of oxygen. As it is an oxidizer, fires involving AN cannot be extinguished by conventional firefighting methods that "smother" a fire by excluding oxygen (air). When heated it will melt, decompose and release toxic gases including nitric acid vapor, nitrogen oxides (NOx) and ammonia gas (NH3). These gases can recombine as they cool, forming a white cloud of AN fumes that is both confusing and a visibility issue for emergency responders. When heated excessively (e.g. as in a fire) it can cause an explosion in an enclosed space and closed containers or vessels may rupture violently. **Firefighting Instructions:** Use water spray or fog for cooling exposed containers. Exercise caution when fighting any chemical fire. Prevent fire-fighting water from entering environment.

**Protection During Firefighting:** Do not enter fire area without proper protective equipment, including respiratory protection.

#### **SECTION 6 - ACCIDENTAL RELEASE MEASURES**

### Personal Precautions, Protective Equipment and Emergency Procedures

**General Measures:** Handle in accordance with good industrial hygiene and safety practice. Avoid breathing (vapor, mist, spray, gas). Do not get in eyes, on skin, or on clothing. Keep away from combustible material.

For Non-Emergency Personnel

Emergency Procedures: Evacuate unnecessary personnel.

For Emergency Personnel

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Protective Equipment: Equip cleanup crew with proper protection. Use appropriate personal protection equipment

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(PPE).

Emergency Procedures: Ventilate area.

#### **Environmental Precautions**

Prevent entry to sewers and public waters. Notify authorities if liquid enters sewers or public waters.

#### Methods and Material for Containment and Cleaning Up

**For Containment:** Dike and contain spill. Notify authorities in accordance with emergency response procedures. Only personnel trained in emergency response should respond. Contain with inorganic absorbents – do not use combustible material such as sawdust or cellulosic material.

**Methods for Cleaning Up:** Soak up spills with inert solids, such as clay or diatomaceous earth as soon as possible. Collect spillage. Store away from other materials.

#### **Reference to Other Sections**

See heading 8, Exposure Controls and Personal Protection

#### **SECTION 7 - HANDLING AND STORAGE**

#### **Precautions for Safe Handling**

Wash hands and other exposed areas with mild soap and water before eating, drinking, or smoking and when leaving work. Provide good ventilation in process area to prevent formation of vapor.

#### Conditions for Safe Storage, Including Any Incompatibilities

**Storage Conditions:** Keep only in a cool, well ventilated place away from: Keep container closed when not in use. **Incompatible Materials:** Strong bases, strong acids. copper, copper bearing alloys, mild steels, wood products, chlorides, chrome.

### **SECTION 8 - EXPOSURE CONTROLS/PERSONAL PROTECTION**

#### **Control Parameters**

No additional information available.

#### **Exposure Controls**

**Appropriate Engineering Controls:** Ensure all national/local regulations are observed. Ensure adequate ventilation, especially in confined areas.

Personal Protective Equipment: Avoid all unnecessary exposure.



**Appearance** 





Materials for Protective Clothing: Heat and chemical resistant materials and fabrics advised.

Hand Protection: Wear protective gloves.

Eve Protection: Chemical goggles or safety glasses.

### **SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES**

Information on Basic Physical and Chemical Properties

Physical State : Liquid

Odor : Characteristic
Odor Threshold : Not available

**pH** : 4.5-5

Evaporation Rate: Not availableMelting Point: Not availableFreezing Point: Not availableBoiling Point: Not available

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Clear, colorless liquid

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Flash Point Not available **Auto-ignition Temperature** Not available **Decomposition Temperature** Not available Flammability (solid, gas) Not available **Lower Flammable Limit** Not available **Upper Flammable Limit** Not available **Vapor Pressure** : Not available Relative Vapor Density at 20 °C : Not available **Relative Density** Not available

Specific gravity / density : Not available Specific Gravity : Not available : Not available

**Solubility** : Water: Solubility in water of component(s) of the mixture: ammonium

nitrate: 190 g/100ml

Partition Coefficient: N-Octanol/Water : Not available Viscosity : Not available

**Explosion Data – Sensitivity to Mechanical**: Not sensitive to mechanical impact. Protect material from drying out.

Impact

Explosion Data – Sensitivity to Static : Not sensitive to static discharge.

**Discharge** 

#### **SECTION 10 - STABILITY AND REACTIVITY**

**Reactivity**: Hazardous reactions are unlikely to occur under normal circumstances. High water content significantly reduces or eliminates the ability of ammonium nitrate to intensify fire, accelerate the burning of other combustible materials or to quickly or violently react as an oxidizer

Chemical Stability: Stable under normal conditions.

Possibility of Hazardous Reactions: Hazardous polymerization will not occur.

Conditions to Avoid: Extremely high temperatures and/or evaporating off excessive water. Incompatible materials.

**Incompatible Materials:** Dissolved metals (copper, zinc, iron, cadmium, manganese, nickel, chromium) act as a catalyst for decomposition. The total level of these heavy metal contaminants should be maintained to less than 50 ppm. Copper should be maintained to less than 1 ppm. Strong acids. Strong bases. Strong oxidizers. Halogens. Chlorine compounds, chlorinated inorganics (potassium, calcium and sodium hypochlorite) and hydrogen peroxides. Combustible materials. Organic materials.

Hazardous Decomposition Products: Nitrogen oxides (NO<sub>X</sub>),. Ammonia (NH<sub>3</sub>). Nitric acid (HNO<sub>3</sub>).

### **SECTION 11 - TOXICOLOGICAL INFORMATION**

Information on Toxicological Effects - Product

Acute Toxicity: Not classified LD50 and LC50 Data: Not available Skin Corrosion/Irritation: Not classified

**pH**: 4.5-5

Serious Eye Damage/Irritation: Causes eye irritation

**pH:** 4.5 - 5

Respiratory or Skin Sensitization: Not classified

Germ Cell Mutagenicity: Not classified

Teratogenicity: Not classified Carcinogenicity: Not classified

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Specific Target Organ Toxicity (Repeated Exposure): Not classified

Reproductive Toxicity: Not classified

Specific Target Organ Toxicity (Single Exposure): Not classified



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Aspiration Hazard: Not classified

Symptoms/Injuries After Inhalation: May cause respiratory irritation. Symptoms/Injuries After Skin Contact: May cause skin irritation. Symptoms/Injuries After Eye Contact: May cause eye irritation.

Symptoms/Injuries After Ingestion: Ingestion is likely to be harmful or have adverse effects.

Chronic Symptoms: None known.

Additional Information: Extreme thermal burn damage to skin and tissue.

### Information on Toxicological Effects - Ingredient(s)

LD50 and LC50 Data:

Ammonium nitrate (6484-52-2)	
LD50 Oral Rat	4820 mg/kg
LC50 dermal rabbit	> 3000 mg/kg
ATE US (oral)	4820 mg/kg bodyweight

#### **SECTION 12: ECOLOGICAL INFORMATION** Toxicity Not classified Persistence and Degradability Ammonium Nitrate solution 20 % Persistence and Degradability Not established. Bioaccumulative Potential Ammonium Nitrate solution 20 % Bioaccumulative Potential Not established. Ammonium nitrate (6484-52-2) BCF fish 1 (no bioaccumulation expected) Log Pow -3.1 (at 25 °C)

Mobility in Soil Not available

**Other Adverse Effects** 

Other Information: Avoid release to the environment.

#### **SECTION 13 - DISPOSAL CONSIDERATIONS**

**Waste Disposal Recommendations:** Dispose in a safe manner in accordance with local/ national regulations. **Additional Information:** Avoid release to the environment.

#### **SECTION 14 - TRANSPORT INFORMATION**

In Accordance with DOT

Not regulated for transport

In Accordance with IATA

No additional information available.

In Accordance with TDG

No additional information available.

### **SECTION 15 - REGULATORY INFORMATION**

**US Federal Regulations** 

Ammonium nitrate (6484-52-2)

Listed on the United States TSCA (Toxic Substances Control Act) inventory

**Canadian Regulations** 

**Ammonium Nitrate solution 20-50%** 

WHMIS Classification No additional information available.

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Ammonium nitrate (6484-52-2)		
Listed on the Canadian DSL (Domestic Substances List)		
WHMIS Classification	No additional information available.	
This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR) and		

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations (CPR) and the SDS contains all of the information required by CPR.

### SECTION 16: OTHER INFORMATION, INCLUDING DATE OF PREPARATION OR LAST REVISION

Revision Date : 07/20/2020 Other Information : None

**GHS Full Text Phrases:** 

Eye Irrit. 2B	Eye irritation Category 2B
H320	May cause eye irritation

#### Party Responsible for the Preparation of This Document

Dyno Nobel Inc.

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